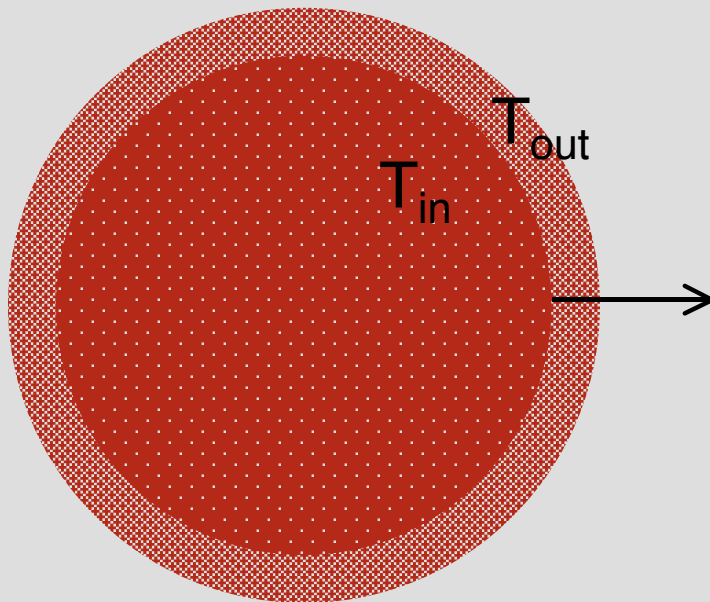


Next, consider an optically thick source:

- Already shown that in the interior, radiation will be described by the Planck function.
- Radiation escaping from the source will be modified because the temperature (and thus the Planck function) **varies** along the path.

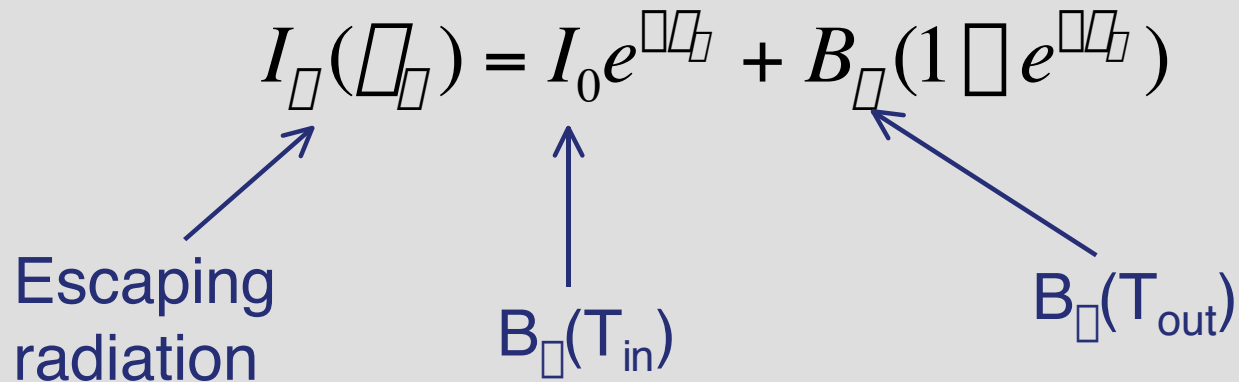
Example: model a star using a two layer model:



Radiation starts from the inner layer as blackbody radiation at temperature T_{in} .

Escapes through an atmosphere of optical depth τ and temperature T_{out} .

Use same solution as before to describe change in intensity of the radiation:

$$I_{\lambda}(\tau_{\lambda}) = I_0 e^{-\tau_{\lambda}} + B_{\lambda}(1 - e^{-\tau_{\lambda}})$$


Escaping radiation

$B_{\lambda}(T_{in})$

$B_{\lambda}(T_{out})$

Valid provided that all the gas is in thermal equilibrium (LTE).

Assume that optical depth of outer layer is small and use approximate expansion for the exponential as before:

$$I_{\lambda}(\tau_{\lambda}) = B_{\lambda}(T_{in})e^{-\tau_{\lambda}} + B_{\lambda}(T_{out})[1 - e^{-\tau_{\lambda}}]$$

$$I_{\lambda}(\tau_{\lambda}) = B_{\lambda}(T_{in})[1 - \tau_{\lambda}] + B_{\lambda}(T_{out})\tau_{\lambda}$$

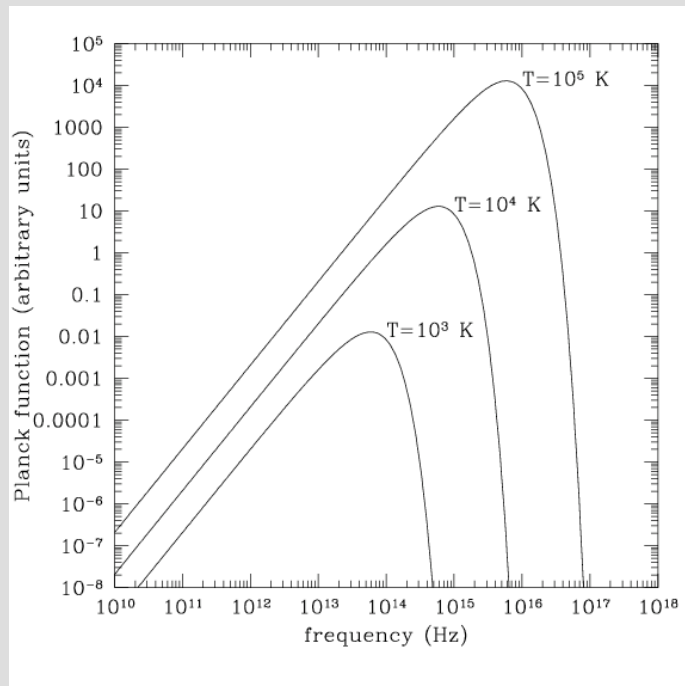
$$I_{\lambda}(\tau_{\lambda}) = B_{\lambda}(T_{in}) + \tau_{\lambda}[B_{\lambda}(T_{out}) - B_{\lambda}(T_{in})]$$

$$I_{\nu}(\nu) = B_{\nu}(T_{in}) + \tau_{\nu} [B_{\nu}(T_{out}) - B_{\nu}(T_{in})]$$

Initial radiation
intensity

Change in intensity caused
by the outer layer. Depends
upon frequency.

Recall that intensity of blackbody radiation increases at **all** frequencies as the temperature goes up.



Sign of the second term depends
upon whether $B_{\nu}(T_{out})$ is larger or
smaller than $B_{\nu}(T_{in})$ - i.e. on
whether $T_{out} > T_{in}$.

$$I_{\nu}(\tau_{\nu}) = B_{\nu}(T_{in}) + \tau_{\nu} [B_{\nu}(T_{out}) - B_{\nu}(T_{in})]$$

1) $T_{out} > T_{in}$: second term is positive:

Escaping intensity is **larger** at frequencies where τ_{ν} is greatest (frequencies corresponding to spectral lines). Expect emission lines on top of the continuum.

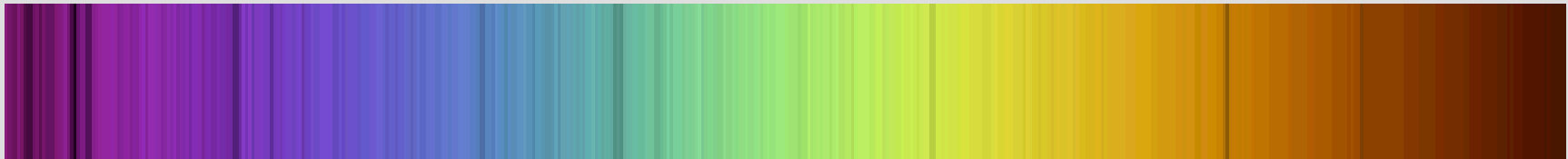
2) $T_{out} < T_{in}$: second term is negative:

Escaping intensity is **reduced** at frequencies where τ_{ν} is greatest (frequencies corresponding to spectral lines). Expect absorption lines superimposed on the continuum.

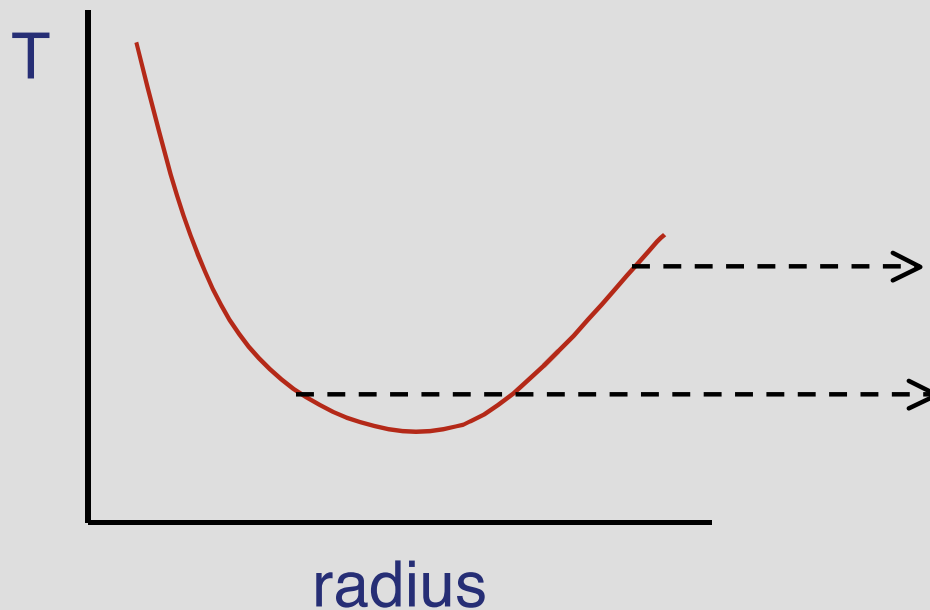
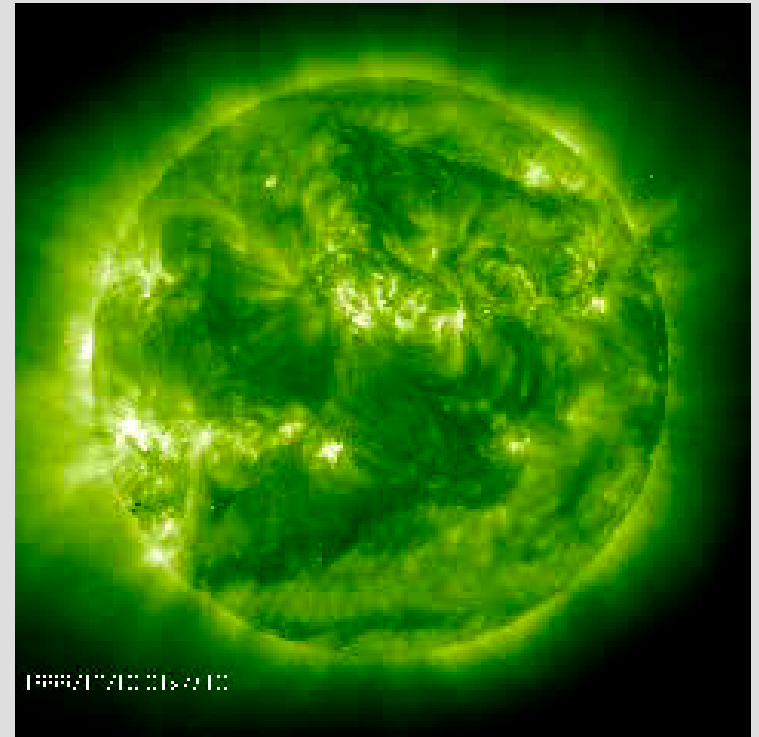
For the Sun, temperature near the optical photosphere **decreases** outward (as it must since energy transport is from the center to the outside).

In second regime: $T_{\text{out}} < T_{\text{in}}$

Expect to see an **absorption spectrum**, as observed:



Note: see strong UV and X-ray emission from the Solar corona, so obviously the temperature there is much hotter than that of the photosphere...



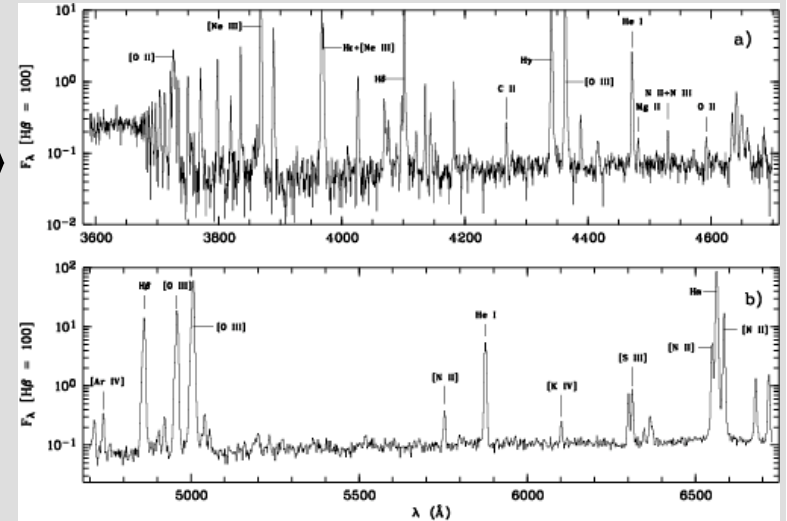
UV radiation comes from region where T increasing, so emission line spectrum

Optical radiation comes from region where T decreasing, so absorption spectrum

Summary:

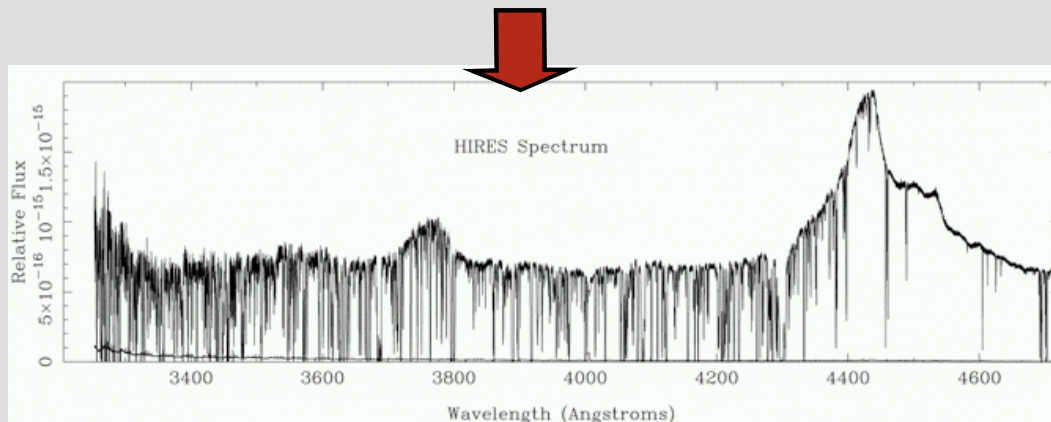
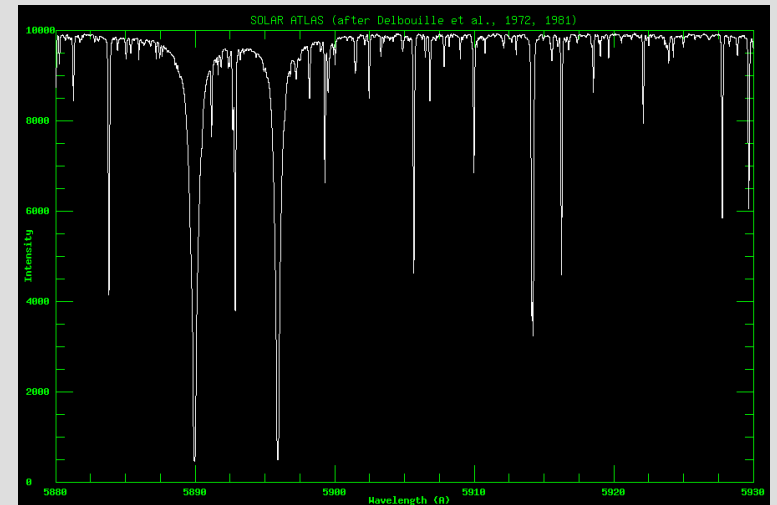
Emission line spectra:

- Optically thin volume of gas with no background light
- Optically thick gas in which T increases outward



Absorption line spectra:

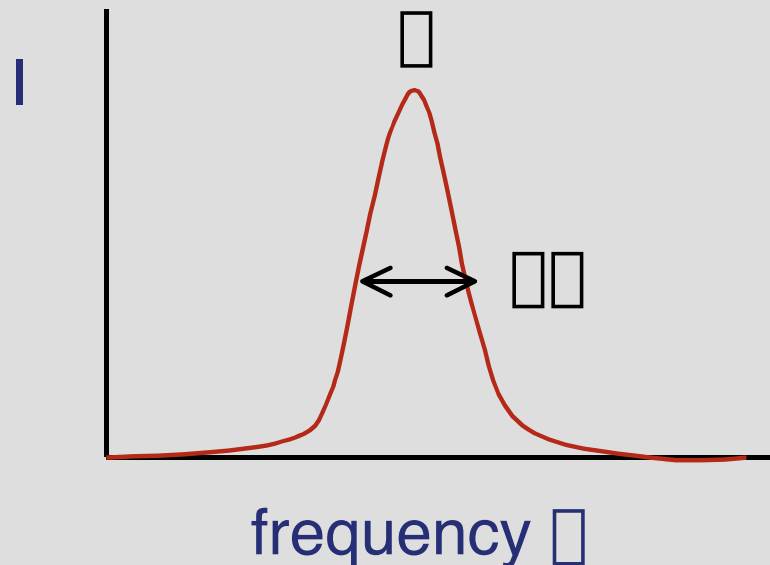
- Cold gas lies in front of a source of radiation at a higher temperature



Broadening of spectral lines

An individual atom making a transition between energy levels emits one photon with a well-defined energy / frequency.

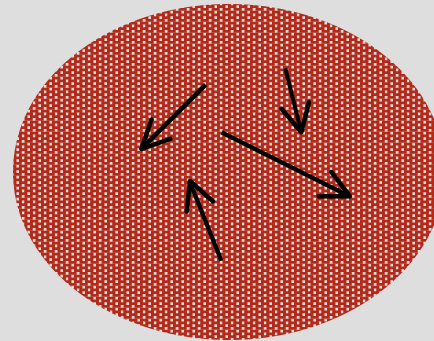
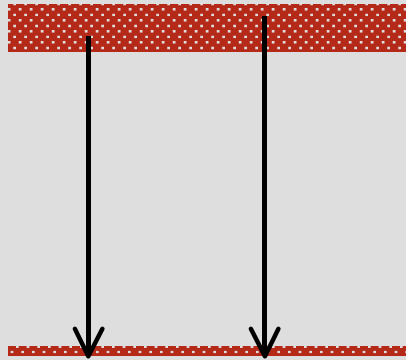
However, profiles of real spectral lines are not infinitely narrow.



e.g. for an emission line, width of the spectral line $\square\square$ could be defined as the full width at half the maximum intensity of the line.

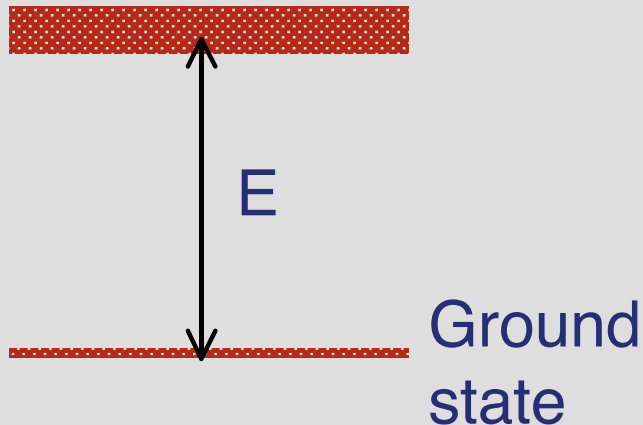
Details of definition don't matter - important to see what causes lines to have finite width.

Two basic mechanisms:



- 1) Energy levels themselves are not infinitely sharp:
emitted photons have a range of frequencies
- 2) Atoms in the gas are moving relative to the observer:
observed photons don't have the same frequency as the emitted photons because of the Doppler effect.

Natural linewidth

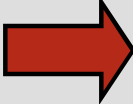


Consider excited state with energy E above the ground state.

Electrons in excited state remain there for average time τ before decaying to ground state.

Uncertainty principle: energy difference between states is uncertain by an amount ΔE given by:

$$\Delta E \Delta t \approx \frac{h}{2\pi}$$

But since $E = h\nu$, $\Delta E = h\Delta\nu$  $\Delta\nu \approx \frac{1}{2\pi\tau}$

Broadening due to this effect is called the natural linewidth.

Natural linewidth sets absolute minimum width of spectral lines. However, normally very small - other effects dominate.

e.g. for hydrogen $n=2$ to $n=1$ transition (Lyman α transition) the lifetime is of the order of 10^{-9} s.

Natural linewidth is $\sim 10^8$ Hz.

Compare to frequency of transition: $\frac{\nu}{\nu} \approx 10^{17}$

In **astrophysical** situations, other processes will often give much larger linewidths than this.

Collisional broadening

In a dense gas, atoms are colliding frequently. This effectively reduces the lifetime of states further, to a value smaller than the quantum mechanical lifetime.

If the frequency of collisions is ν_{col} , then expect to get a collisional linewidth of about $\propto \nu_{\text{col}}$.

Frequency of collisions increases with density - expect to see broader lines in high density regions as compared to low density ones.

e.g. a main sequence star (small radius) has a higher density at the photosphere than a giant of the same surface temperature. Spectral lines in the main sequence star will be broader than in the giant.