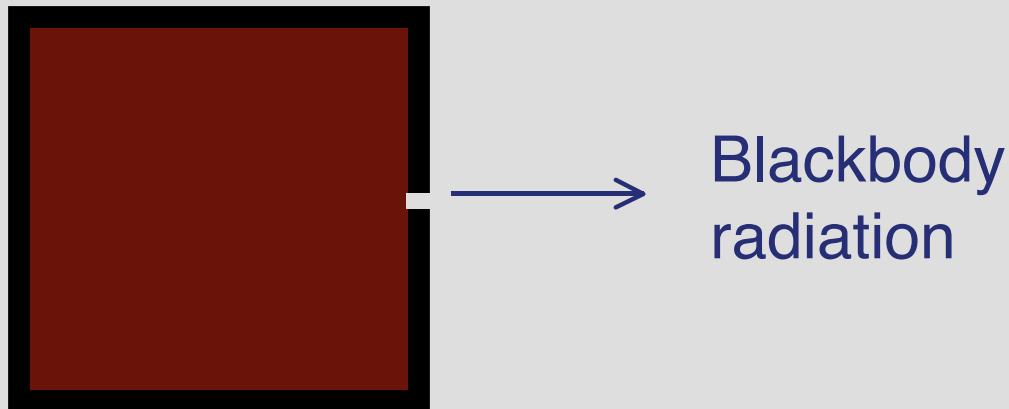


# Sources of radiation

Most important type of radiation is **blackbody radiation**. This is radiation that is in thermal equilibrium with matter at some temperature  $T$ .

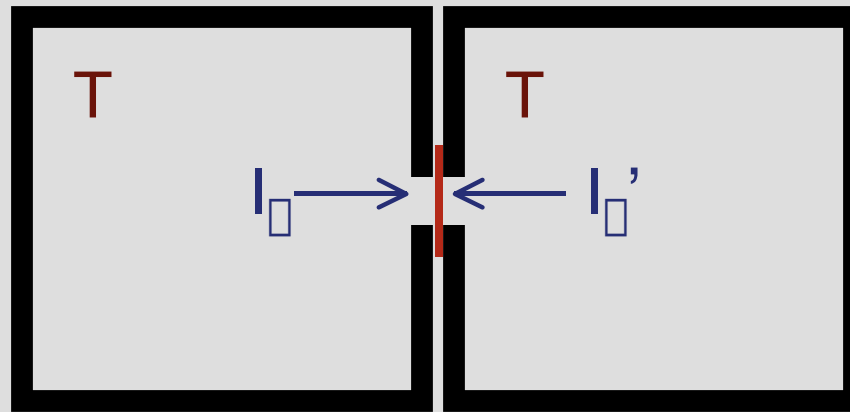
Lab source of blackbody radiation: hot oven with a small hole which does not disturb thermal equilibrium inside:



Important because:

- Interiors of stars (for example) are like this
- Emission from many objects is roughly of this form.

A general principle in physics is that **energy cannot flow spontaneously from a cold body to a hot one** (a statement of the second law of thermodynamics). Implies that the intensity of blackbody radiation can only depend on  $T$ :



Imagine two blackbody cavities at the same temperature  $T$ , separated by a filter that transmits only radiation with frequency close to  $\nu$ . If:

$$I_{\nu} \neq I'_{\nu}$$

...there would be a net energy flow, in violation of the 2nd law. Hence  $I_{\nu}$  can only depend on  $T$ .

## Spectrum of blackbody radiation

The frequency dependence of blackbody radiation is given by the **Planck function**:

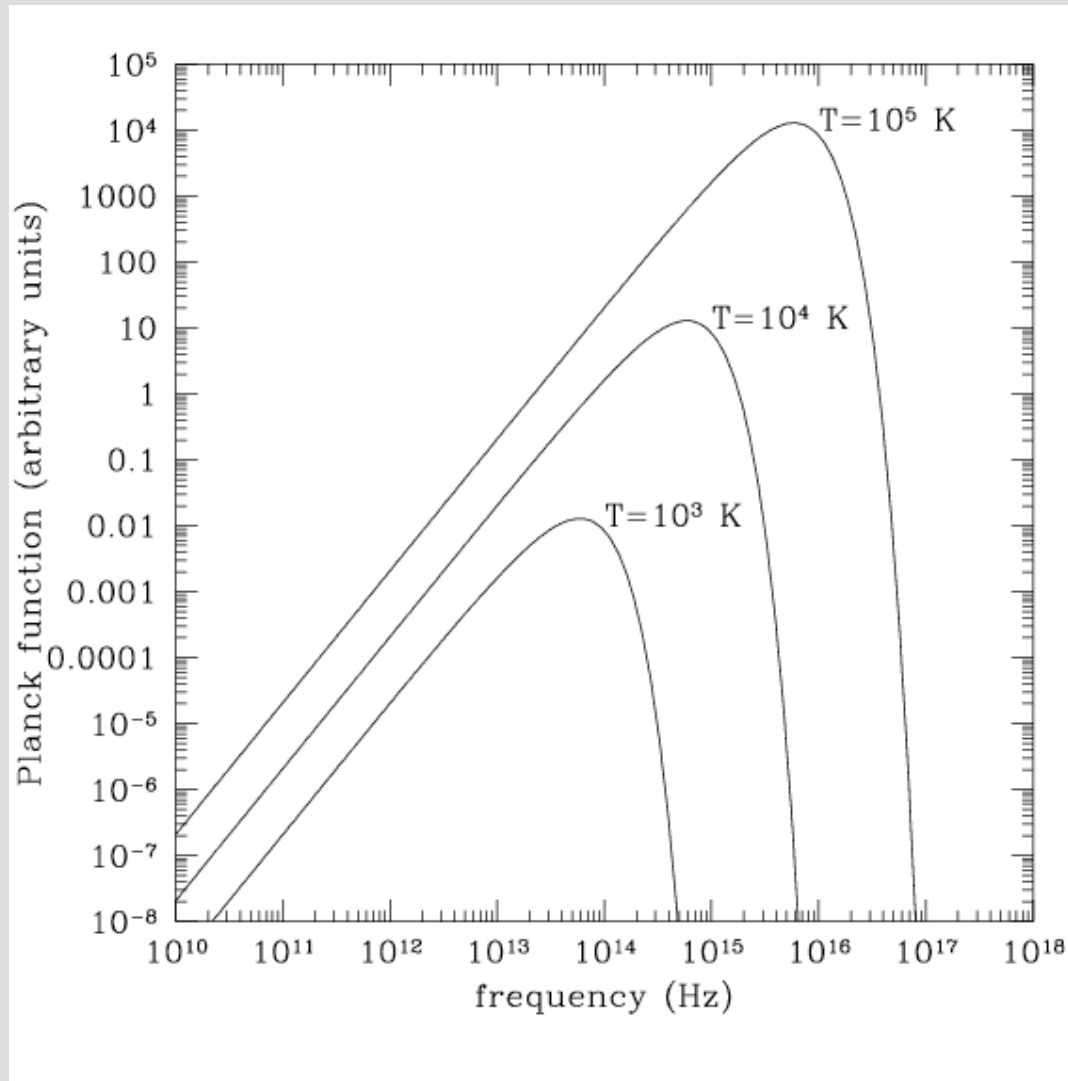
$$B_{\nu}(T) = \frac{2h\nu^3 / c^2}{\exp(h\nu/kT) - 1}$$

- $h = 6.63 \times 10^{-27}$  erg s is Planck's constant
- $k = 1.38 \times 10^{-16}$  erg K<sup>-1</sup> is Boltzmann's constant

Same units as specific intensity: erg s<sup>-1</sup> cm<sup>-2</sup> steradian<sup>-1</sup> Hz<sup>-1</sup>

Derivation in *Harwit*, or see different derivation on the web page.

# Properties of blackbody radiation



Plot  $B_{\nu}(T)$ :

- Continuous spectrum
- Increasing  $T$  increases  $B_{\nu}$  at **all** frequencies
- Higher temperature shifts the peak to higher frequency / shorter wavelength.

**Thermal radiation:**

$$S_{\nu} = B_{\nu}$$

**Blackbody radiation:**

$$I_{\nu} = B_{\nu}$$

Differentiate  $B_{\nu}(T)$  with respect to frequency and set resulting expression to zero to find where the Planck function peaks.

$$h\nu_{\max} = 2.82kT$$

$$\nu_{\max} = 5.88 \times 10^{10} T \text{ Hz K}^{-1}$$

*Wien displacement law* - peak shifts linearly with increasing temperature to higher frequency.

*Rayleigh-Jeans law*: for low frequencies  $h\nu \ll kT$ :

$$B_{\nu}^{RJ}(T) = \frac{2\nu^2}{c^2} kT$$

Often valid in the radio part of the spectrum, at frequencies far below the peak of the Planck function.

The energy density of blackbody radiation:

$$u(T) = aT^4$$

...where  $a = 7.56 \times 10^{-15} \text{ erg cm}^{-3} \text{ K}^{-4}$  is the radiation constant.

The emergent flux from a surface emitting blackbody radiation is:

$$F = \sigma T^4$$

...where  $\sigma = 5.67 \times 10^{-5} \text{ erg cm}^{-2} \text{ K}^{-4} \text{ s}^{-1}$  is the Stefan-Boltzmann constant.

e.g. star, radius  $R$ , temperature  $T$ , emitting as a blackbody, has a luminosity:

$$L = 4\pi R^2 \sigma T^4$$

Emission from most astronomical sources is only roughly described by the Planck function (if at all).

Source has emergent flux  $F$  (integrated over all frequencies), define the **effective temperature**  $T_e$  via:

$$F \equiv \sigma T_e^4$$

Effective temperature is the temperature of a blackbody that emits same flux. e.g. for the Sun:

$$L_{sun} = 4\pi R_{sun}^2 \sigma T_e^4$$

...find  $T_e = 5770$  K.

Note: effective temperature is well-defined even if the spectrum is nothing like a blackbody.

## Which objects have blackbody spectra?

Radiation will be blackbody radiation wherever we have matter in thermal equilibrium with radiation - i.e. at large optical depth. To show this, consider transfer equation with  $S_\lambda = B_\lambda(T)$ , and assume  $T$  is constant:

$$\frac{dI_\lambda}{d\lambda} = -\kappa I_\lambda + B_\lambda(T)$$

Can integrate this equation, with solution:

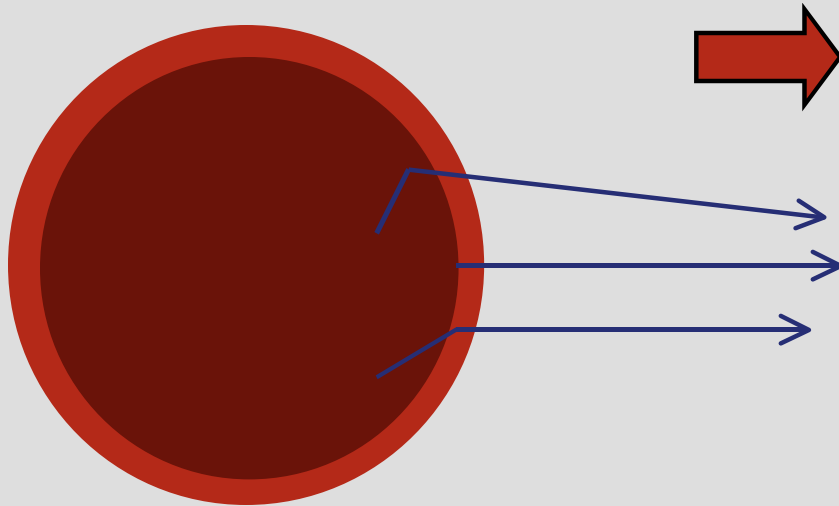
$$I_\lambda(\lambda) = B_\lambda + \underbrace{e^{-\kappa\lambda} [I_0 - B_\lambda]}_{\rightarrow 0 \text{ as } \lambda \text{ becomes large}}$$

$\rightarrow 0$  as  $\lambda$  becomes large

where  $I_0$  is the value of  $I_\lambda$  at  $\lambda = 0$ . Conclude  $I_\lambda = B_\lambda$  at high optical depth, e.g. in the center of a star.



Recall interpretation of optical depth: at  $\tau = 1$  there is (very roughly speaking) a 50% chance that a photon headed toward us will suffer an absorption or scattering along the way.

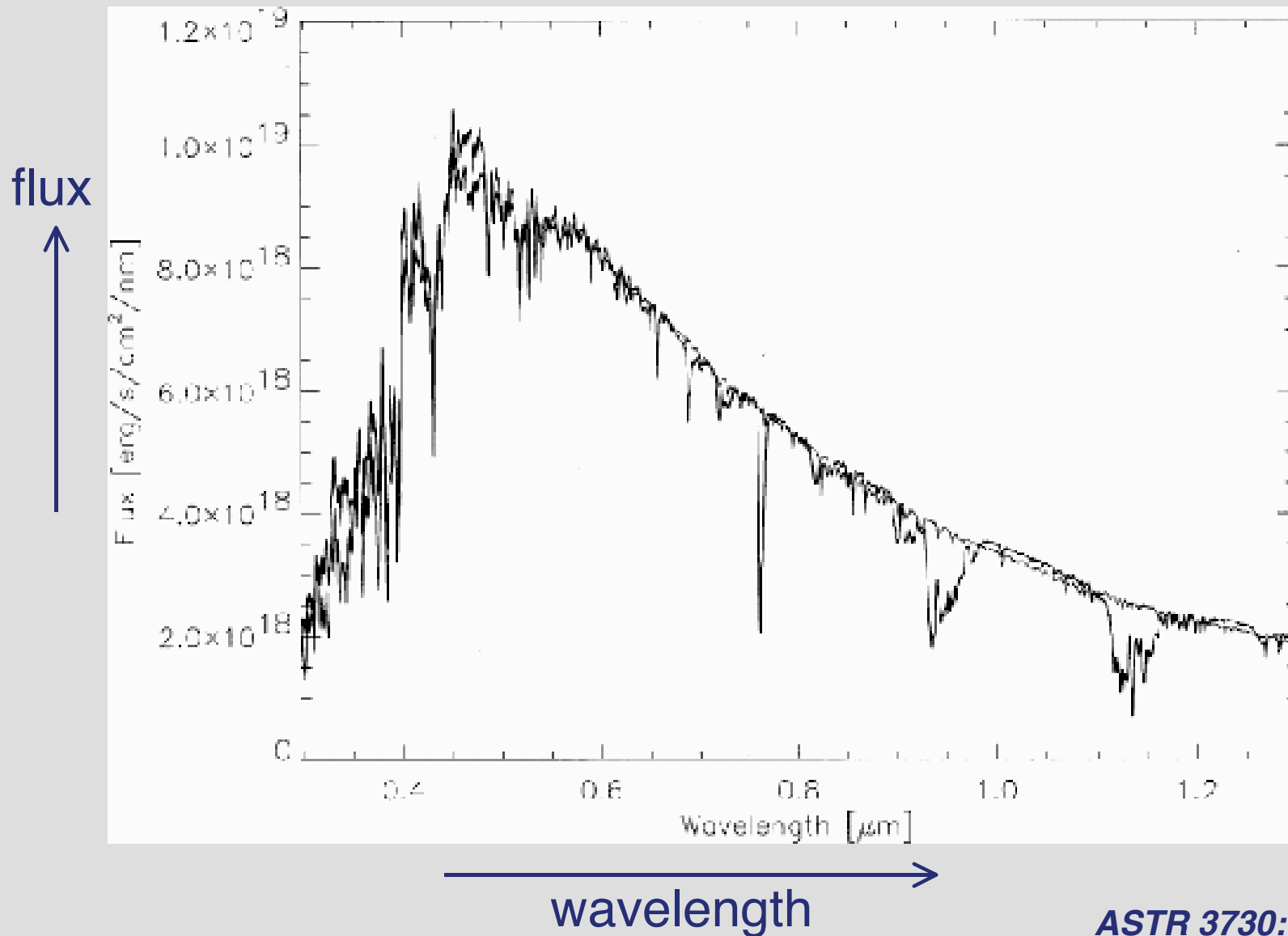


By definition, photons we can observe from an optically thick source come from near the  $\tau = 1$  surface.

Since  $\tau = 1$  is *not* 'high optical depth', observed radiation from an optically thick source is not necessarily blackbody spectrum.

Deviations will depend upon the frequency dependence of the opacity around the  $\tau = 1$  surface (the photosphere).

Solar spectrum - approximately of blackbody form. Very cool stars show larger departures from thermal spectra.



## Giant planets

Two sources of radiation:

- Directly reflected Sun light
- Absorbed Solar radiation, reradiated as a cool blackbody

e.g. Jupiter:  $L_{sun} = 3.86 \times 10^{33} \text{ erg s}^{-1}$

$a_J = 7.8 \times 10^{13} \text{ cm}$       Jupiter orbital radius

$R_J = 7.1 \times 10^9 \text{ cm}$       Jupiter radius

Solar radiation incident on the planet is:

$$L_J = \frac{\pi R_J^2}{4\pi a_J^2} L_{sun} \approx 2 \times 10^9 L_{sun}$$

Suppose planet directly reflects 10% - in the optical Jupiter is  $\sim 10^{10}$  times fainter than the Sun as seen from another star - about 25 magnitudes.

Absorb and reradiate as a blackbody:

$$L_J = 4\pi R_J^2 \sigma T_J^4$$

If all Sun light absorbed, estimate  $T = 120$  K. Use:

$$h\nu_{\max} = 2.82kT$$

Find  $\nu_{\max} = 7 \times 10^{12}$  Hz, which corresponds to a wavelength of around 40 microns.